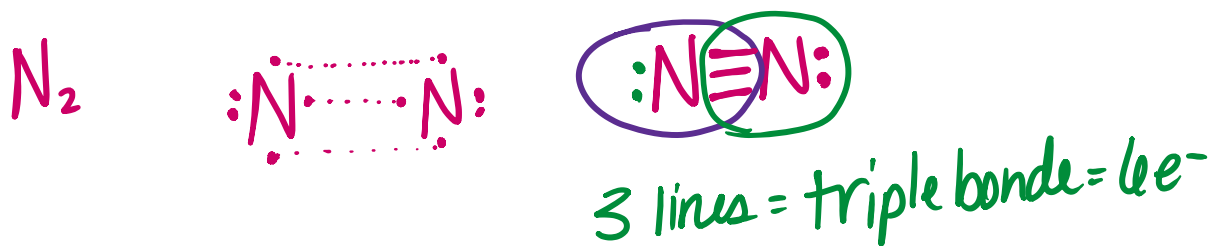
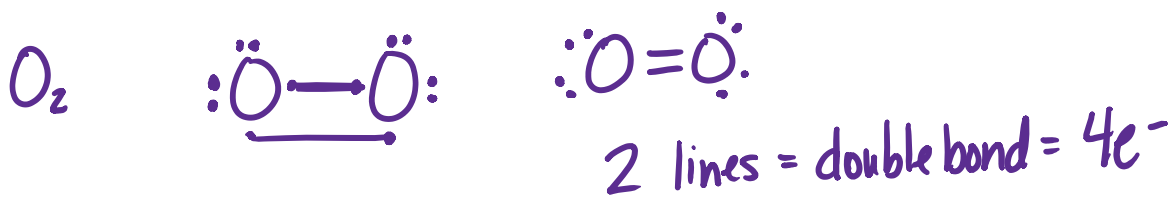


Covalent Bonding - Share valence electrons to get to an octet.  
 (duet for hydrogen)

Lewis Dot

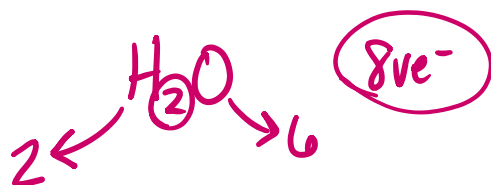


Shared electrons can be counted as owned by both atoms



How to!

1. Count all ve
2. Determine central atom  
 (usually, its the element there is only 1 of)



3. Draw single bond from central atom
4. Put all remaining ve<sup>-</sup> on atoms as lone pairs (..)

5. Turn lone pairs into double or triple bonds to give every atom an octet.

